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IMPORTANCE OF USING THE WATER CIRCULATION SYSTEM IN THE MINING ENRICHMENT PLANTS

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Teghut tailings as a representative of the general condition and risks of tailings in Armenia are studied accordingly, we have made laboratory (solid material and water composition) and social analysis and we have concluded that the high amount of hazardous heavy metals in these tailings spread to nearby territories and contaminate the soil, water and air, enter the food chain, and adversely affect the human health. At first glance, these problems seem unsolvable. However, correction measures must be implemented in order to reduce the pollution.

Keywords: mine tailing, water circulation, heavy metals, metallic ores, tails enrichment.

Introduction. Mine tailings and accidents related to them are one of the most dangerous environmental problems all over the world. This problem is specially acute in small countries where mining industries are important, particularly mining of metallic ores. Armenia, being a mountainous country with a small area, problems of occupation of space and soil and water pollution become particularly acute and urgent to be solved. In fact, there are more than 1 million tons of mining wastes in Republic of Armenia already located in the tailing dams of more than 20 enrichment tailings [1]. The construction of new storage facilities, and consequently, the occupation of new territories appear regularly, and it becomes a serious, unsolvable problem since those areas used for tailings become unsuitable for further use.

Conducting experiments and discussion of results. Besides the destruction of ancient forest ecosystem, the Teghout mine also causes other serious negative consequences, like the extreme deterioration of the Shnogh River water flowing in the areas adjacent to the mine [2].

The quality of the river water is steadily deteriorating parallelly with the mining activity from 2009-2015. As compared with 2009, the average concentrations of some elements (Mn, Ni, Cu, Zn, As, Se, Mo, Pb) have increased according to the hydrological seasons. As a consequence, the Shnogh River is classified as a risk water body according to the EU Water Framework Directive [3].

There are sufficient grounds to suspect that the tails produced by the ore dressing plant flow directly into the river, whose water is used to irrigate the orchards. River water is used for irrigation and as drinking water for domestic animals. Local farmers have reported health problems in livestock and the decreasing quality in their crops [4].

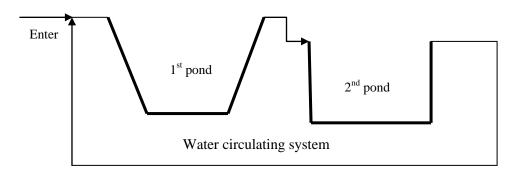


Fig. Technical scheme of Teghut tailing

The Teghut tailing consists of 2 parts: the first is a genuine tailing dump, where the mixture from the mine is emptied (stream 1) and undergoes sedimentation. After this process, the effluent is transferred to the second part, a pond where the water is cleaned by sedimentation (Stream 2) and is pumped back to the factory(fig.).

Water samples from the first and second ponds of the tailings were collected on February 2015. Each sample was split into two parts, and one of them was filtered through a $0.45\mu m$ filter. The retentate (Table 3) and filtrate (Table 4) obtained were analyzed along with the other part of the samples using ICP-MS. The results are presented in Table 1 for the first pond, and Table 2 for the second pond.

It must be noted that the sedimentation and removal of suspended particles in the first pond reduces the levels of several metals, including Ag, Mo, Nb, Rb, Zn, Cu, Fe, Ca, K, Si, Cl.

Concentration of several elements in the water collected from the first (Stream 1) and from the second (Stream 2) ponds of the Teghut tailing

Table 1

Table 2

Element	Concentration, ppb	Std.Error	I	Element	Concentration, ppb	Std.Error
Ba	<lod*< td=""><td>218.632</td><td></td><td>Ba</td><td><lod*< td=""><td>153.291</td></lod*<></td></lod*<>	218.632		Ba	<lod*< td=""><td>153.291</td></lod*<>	153.291
Sb	<lod< td=""><td>62.002</td><td></td><td>Sb</td><td><lod< td=""><td>51.768</td></lod<></td></lod<>	62.002		Sb	<lod< td=""><td>51.768</td></lod<>	51.768
Sn	<lod< td=""><td>64.834</td><td></td><td>Sn</td><td><lod< td=""><td>42.046</td></lod<></td></lod<>	64.834		Sn	<lod< td=""><td>42.046</td></lod<>	42.046
Cd	34.289	17.805		Cd	<lod< td=""><td>35.339</td></lod<>	35.339
Pd	17.528	10.674		Pd	<lod< td=""><td>17.701</td></lod<>	17.701
Ag	79,304	16.844		Ag	<lod< td=""><td>17.979</td></lod<>	17.979
Mo	61.688	35.327		Mo	585.857	22.427
Nb	1203.873	4.965		Nb	14.774	5.176
Zr	12.876	12.325		Zr	<lod< td=""><td>16.792</td></lod<>	16.792
Sr	<lod< td=""><td>25.672</td><td></td><td>Sr</td><td>838.346</td><td>32.181</td></lod<>	25.672		Sr	838.346	32.181
Rb	641.833	4.905		Rb	14.970	4.244
Bi	26.517	15.162		Bi	<lod< td=""><td>32.006</td></lod<>	32.006
As	24.342	24.005		As	<lod< td=""><td>15.5</td></lod<>	15.5
Se	<lod< td=""><td>25.261</td><td></td><td>Se</td><td><lod< td=""><td>28.171</td></lod<></td></lod<>	25.261		Se	<lod< td=""><td>28.171</td></lod<>	28.171
Au	<lod< td=""><td>33.695</td><td></td><td>Au</td><td><lod< td=""><td>29.984</td></lod<></td></lod<>	33.695		Au	<lod< td=""><td>29.984</td></lod<>	29.984
Pb	<lod< td=""><td>22.673</td><td></td><td>Pb</td><td><lod< td=""><td>24.957</td></lod<></td></lod<>	22.673		Pb	<lod< td=""><td>24.957</td></lod<>	24.957
W	201.852	131.103		W	295.918	138.622
Zn	79.733	32.411		Zn	<lod< td=""><td>45.364</td></lod<>	45.364
Cu	79.884	44.12		Cu	<lod< td=""><td>92.226</td></lod<>	92.226
Ni	<lod< td=""><td>87.706</td><td></td><td>Ni</td><td><lod< td=""><td>86.177</td></lod<></td></lod<>	87.706		Ni	<lod< td=""><td>86.177</td></lod<>	86.177
Со	<lod< td=""><td>120.597</td><td></td><td>Co</td><td><lod< td=""><td>95.784</td></lod<></td></lod<>	120.597		Co	<lod< td=""><td>95.784</td></lod<>	95.784
Fe	2208.537	256.969		Fe	<lod< td=""><td>176.899</td></lod<>	176.899
Mn	<lod< td=""><td>200.6</td><td></td><td>Mn</td><td><lod< td=""><td>220.968</td></lod<></td></lod<>	200.6		Mn	<lod< td=""><td>220.968</td></lod<>	220.968
Cr	<lod< td=""><td>103.499</td><td></td><td>Cr</td><td><lod< td=""><td>165.335</td></lod<></td></lod<>	103.499		Cr	<lod< td=""><td>165.335</td></lod<>	165.335
V	<lod< td=""><td>122.384</td><td></td><td>V</td><td>91.796</td><td>56.418</td></lod<>	122.384		V	91.796	56.418
Ti	<lod< td=""><td>176.665</td><td></td><td>Ti</td><td><lod< td=""><td>230.280</td></lod<></td></lod<>	176.665		Ti	<lod< td=""><td>230.280</td></lod<>	230.280
Ca	235397.172	3919.392		Ca	371721.8	5556.849
K	14980.718	489.775		K	9951.237	388.063
Al	<lod< td=""><td>1651.7</td><td></td><td>Al</td><td><lod< td=""><td>1814.552</td></lod<></td></lod<>	1651.7		Al	<lod< td=""><td>1814.552</td></lod<>	1814.552
Р	<lod< td=""><td>5521.768</td><td></td><td>Р</td><td><lod< td=""><td>453.159</td></lod<></td></lod<>	5521.768		Р	<lod< td=""><td>453.159</td></lod<>	453.159
Si	8702.750	534.051		Si	6465.233	482.158
Cl	19085.695	304.394		Cl	16767.1	249.843

*<LOD= below the detection limit

	Table	3
Sediment from	stream	1

Element	Concentration,	Std.Error	
	ppb		
Ba	165.06	76.823	
Sb	<lod*< td=""><td>30.055</td></lod*<>	30.055	
Sn	<lod< td=""><td>28.762</td></lod<>	28.762	
Cd	<lod< td=""><td>16.06</td></lod<>	16.06	
Pd	<lod< td=""><td>7.650</td></lod<>	7.650	
Ag	<lod< td=""><td>7.785</td></lod<>	7.785	
Мо	26.289	3.609	
Nb	<lod< td=""><td colspan="2">3.683</td></lod<>	3.683	
Zr	64.801	5.329	
Sr	263.562	9.212	
Rb	13.636	2.253	
Bi	<lod< td=""><td>8.904</td></lod<>	8.904	
As	<lod< td=""><td>6.769</td></lod<>	6.769	
Se	<lod< td=""><td>4.142</td></lod<>	4.142	
Au	<lod< td=""><td>13.274</td></lod<>	13.274	
Pb	<lod< td=""><td>7.839</td></lod<>	7.839	
W	<lod< td=""><td>91.57</td></lod<>	91.57	
Zn	26.089	15.799	
Cu	862.764	56.961	
Ni	<lod< td=""><td>57.836</td></lod<>	57.836	
Со	<lod< td=""><td>184.609</td></lod<>	184.609	
Fe	20821.3	441.656	
Mn	<lod< td=""><td>187.312</td></lod<>	187.312	
Cr	46.265	29.369	
V	97.950	33.161	
Ti	1897.329	73.485	
Ca	6834.123	347.057	
K	9205.72	268.349	
Al	14512.4	701.197	
Р	416.569	148.656	
Si	106325.1	14 46.894	
Cl	320.463	46.894 27.421	

Concentration	of	several	el	ements	in	the
water (after filt	tering	g from s	edi	ment) co	olle	cted
from the first	pona	lof i	the	Teghut	tai	ling
(Stream 1)						

Table 4

Composition	Concentration Ppm	Error	
Ba	<lod< td=""><td colspan="2">170.656</td></lod<>	170.656	
Sb	<lod< td=""><td>48.136</td></lod<>	48.136	
Sn	62.094	38.267	
Cd	<lod< td=""><td>33.698</td></lod<>	33.698	
Pd	<lod< td=""><td>18.846</td></lod<>	18.846	
Ag	<lod< td=""><td>17.872</td></lod<>	17.872	
Bal	373363.5	8520.186	
Mo	439.801	21.455	
Nb	11.433	5.612	
Zr	<lod< td=""><td>16.108</td></lod<>	16.108	
Sr	836.307	36.725	
Rb	12.547	4.587	
Bi	<lod< td=""><td>25.359</td></lod<>	25.359	
As	<lod< td=""><td>31.388</td></lod<>	31.388	
Se	<lod< td=""><td>24.945</td></lod<>	24.945	
Au	<lod< td=""><td>36.571</td></lod<>	36.571	
Pb	<lod< td=""><td>24.035</td></lod<>	24.035	
W	406.539	170.887	
Zn	<lod< td=""><td>77.082</td></lod<>	77.082	
Cu	93.725	54.886	
Ni	<lod< td=""><td>100.029</td></lod<>	100.029	
Co	<lod< td=""><td>95.191</td></lod<>	95.191	
Fe	<lod< td=""><td>284.539</td></lod<>	284.539	
Mn	<lod< td=""><td>241.846</td></lod<>	241.846	
Cr	146.399	94.112	
V	<lod< td=""><td>106.679</td></lod<>	106.679	
Ti	<lod< td=""><td>297.458</td></lod<>	297.458	
Ca	444819.6	7195.210	
K	11074.95	438.167	
Al	<lod< td=""><td>2280.635</td></lod<>	2280.635	
Р	<lod< td=""><td>511.237</td></lod<>	511.237	
Si	6225.832	515.487	

*<LOD= below the detection limit

Conclusion. As we saw from the article we have a huge amount of sedimentation in tailings with useful metals, we propose to open a new part in front of the tailing to extract all useful minerals and reuse it, and after a little part of sedimentation, send it to the tailings. We also propose:

• To use the water circulation method in all tailings, it is very effective for reducing the amount of hazardous materials for the environment.

• To reuse all useful products from ore materials and leave as little waste as possible (nowadays, most of the tailings accumulate millions of tons of useful substances).

• Arrange a good isolation of tailings, especially for the floor, because of bad isolation, heavy metals can penetrate into the ground water.

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ՋՐԻ ՇՐՋԱՆԱՌՈՒ ՀԱՄԱԿԱՐԳԻ ՕԳՏԱԳՈՐԾՄԱՆ ԿԱՐԵՎՈՐՈՒԹՅՈՒՆԸ ԼԵՌՆԱՀԱՐՍՏԱՅՄԱՆ ՖԱԲՐԻԿԱՆԵՐՈՒՄ

Ա.Վ. Թադևոսյան, Տ.Ս. Բաղդասարյան, Հ.Ա. Ջարգարյան, Ա.Ա. Սաֆարյան

Ուսումնասիրվել և պոչամբարների պինդ ու հեղուկ նյութերի (ֆազերի) լաբորատոր և դրանց սոցիալ-տնտեսական հետազոտությունների վերլուծության արդյունքում կատարվել է եզրակացություն։ Պոչամբարներից մեծ քանակությամբ վտանգավոր ծանր մետաղներ տարածվում են հարակից տարածքներ և աղտոտում հողը, ջուրը, օդը, ներթափանցում սննդի շղթա և բացասաբար են ազդում մարդու առողջության վրա։ Առաջին հայացքից այս խնդիրները անլուծելի են թվում։ Սակայն մեր կողմից առաջարկվող միջոցառումների շնորհիվ կարելի է աղտոտումը նվազեցնել։ Հայաստանում պոչամբարների ընդհանուր վիճակի և դրանց ռիսկերի վերլուծությունը, ինչպես և ընդունված է, կատարվել է Թեղուտի պոչամբարի օրինակով։

Առանցքային բառեր. պոչամբար, ջրի շրջանառություն, ծանր մետաղներ, մետաղական հանքաքարեր, պոչերի հարստացում։

ВАЖНОСТЬ ИСПОЛЬЗОВАНИЯ СИСТЕМЫ ЦИРКУЛЯЦИИ ВОДЫ В ГОРНО-ОБОГАТИТЕЛЬНЫХ КОМБИНАТАХ

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Рассматривается важность использования системы циркуляционной воды в горнообогатительных комбинатах. С целью общего состояния и риска хвостохранилищ в Армении используется хвостохранилище Техута. Результаты лабораторных (твердый материал и состав воды) и социальных анализов показали, что большое количество опасных тяжелых металлов в этих хвостохранилищах распространилось на близлежащие территории, загрязняя почву, воду и воздух, входят в пищевую цепь и неблагоприятно влияют на здоровье человечества. На первый взгляд, эти проблемы кажутся неразрешимыми. Предлагается ряд мер по коррекции опасных тяжелых металлов с целью уменьшения загрязнения окружающей среды.

Ключевые слова: хвостохранилище, циркуляция воды, тяжелые металлы, металлические руды, обогащение хвостов.